



## Directions in QPPR development to complement the predictive models used in risk assessment of nanomaterials

### Citation

Quik, J. T. K., Bakker, M., van de Meent, D., Poikkimäki, M., Dal Maso, M., & Peijnenburg, W. (2018). Directions in QPPR development to complement the predictive models used in risk assessment of nanomaterials. *NanoImpact*, 11, 58-66. <https://doi.org/10.1016/j.impact.2018.02.003>

### Year

2018

### Version

Peer reviewed version (post-print)

### Link to publication

[TUTCRIS Portal \(http://www.tut.fi/tutcris\)](http://www.tut.fi/tutcris)

### Published in

NanoImpact

### DOI

[10.1016/j.impact.2018.02.003](https://doi.org/10.1016/j.impact.2018.02.003)

### Copyright

© 2018. This manuscript version is made available under the CC-BY-NC-ND 4.0 license <http://creativecommons.org/licenses/by-nc-nd/4.0/>

### License

CC BY-NC-ND

### Take down policy

If you believe that this document breaches copyright, please contact [cris.tau@tuni.fi](mailto:cris.tau@tuni.fi), and we will remove access to the work immediately and investigate your claim.

# Directions in QPPR development to complement the predictive models used in risk assessment of nanomaterials

---

Authors:

Joris T.K. Quik<sup>1</sup>, Martine Bakker<sup>2</sup>, Dik van de Meent<sup>3</sup>, Mikko Poikkimäki<sup>4</sup>, Miikka Dal Maso<sup>4</sup> and Willie Peijnenburg<sup>2,5</sup>

<sup>1</sup> Centre for Sustainability, Environment and Health, National Institute of Public Health and the Environment (RIVM), Bilthoven, 3720BA Bilthoven, The Netherlands

<sup>2</sup> Centre for Safety of Substances and Products, National Institute of Public Health and the Environment (RIVM), Bilthoven, 3720BA Bilthoven, The Netherlands

<sup>3</sup> Institute for Water and Wetland Research, Department of Environmental Science, Radboud University Nijmegen, P.O. Box 9010, NL-6500 GL Nijmegen

<sup>4</sup> Aerosol Physics, Laboratory of Physics, Faculty of Natural Sciences, Tampere University of Technology, P.O. Box 692, FI-33101 Tampere, Finland

<sup>5</sup> Institute of Environmental Sciences, Leiden University, 2300 RA Leiden, The Netherlands

Words: 6424

Tables: 5

Figures: 1

Corresponding author: Joris T.K. Quik - [joris.quik@rivm.nl](mailto:joris.quik@rivm.nl)

25 **Abstract**

26

27 There is an increasing need for predictive risk assessment of nanomaterials (NMs) using methods  
28 that are rapid, accurate and resource efficient. To fulfill this need, the development and use of  
29 Quantitative Property Property Relationships (QPPRs) for estimating the hazard of NMs and NM-  
30 related parameters in exposure modelling seems eminent. In this study, we analyze a selection of  
31 models used for hazard and/or exposure assessment of NMs. This analysis was done by identifying all  
32 the NM-related properties used in these models related to three categories of data: (i) Intrinsic  
33 properties specific to the NM, matrix or experimental conditions, (ii) Extrinsic NM properties related  
34 to interaction between the intrinsic properties and (iii) Measured hazard or exposure data. This  
35 analysis is combined with the current state of QPPR development to recommend further  
36 development of QPPRs for predictive risk assessment of NMs. In particular, the use of descriptors  
37 related to the interaction between a NM and its surroundings, e.g. the attachment efficiency is  
38 proposed.

39 **Key words:** nanomaterial, modelling, in silico, QPPR, QNAR, risk assessment

## 40 1 Introduction

41 With the increasing rate of new nanomaterials (NMs) being developed and applied, an increase in  
42 knowledge gaps is expected for assessing the hazard, exposure and risk of NMs to the environment  
43 and to human health. NMs are expected to be applied in a vast number of variations in e.g. size,  
44 shape, coating and chemical composition. It is not feasible to generate information for every  
45 nanomaterial on the routes of exposure and uptake, and potential bioaccumulation in biota and in  
46 the human body. In addition, generating information on the main interactions with biological  
47 systems, requiring animal testing, may be regarded as unethical in terms of animals use and wasteful  
48 in terms of resource use (Russell and Burch, 1959). Therefore, it is important to develop *in silico*  
49 approaches to aid in the prediction of NM safety based on their physico-chemical properties. *In silico*  
50 methods traditionally refer to the application of computational modeling techniques for predicting  
51 the activity or effects of a chemical based on its chemical structure (Reisfeld and Mayeno, 2012). This  
52 includes Quantitative Structure-Activity Relationships (QSAR) which are more widely used in  
53 pharmacology (Dearden, 2003; Fujita and Winkler, 2016), and are already finding application in the  
54 safety regulation of molecular or ionic substances (European Commission, 2006; ECHA, 2008). QSARs  
55 have already been successfully used in relating structural characteristics to chemical properties and  
56 biological effects of molecular substances in order to fill data gaps (Chen et al., 2014; Singh et al.,  
57 2014; Modarresi et al., 2007). According to REACH, data derived from QSARs may support the  
58 waiving of laboratory testing or serve as a trigger for proposing further testing or used instead of  
59 testing data when certain required conditions are met (ECHA, 2008). A well-known example in this  
60 respect is the *in silico* approach used in the exposure models that are included in REACH, allowing to  
61 predict the solids-water partition coefficient on the basis of the octanol-water partitioning coefficient  
62 ( $K_{ow}$ ) of organic compounds in combination with the fraction organic matter ( $f_{oc}$ ) of solids (Sabljic et  
63 al., 1995). Similarly, the  $K_{ow}$  can be used as a descriptor to calculate the acute toxicity (LC50) of  
64 certain compounds for mice (Dearden, 2003).

65 These relationships, although strictly not addressing 'activity' in the pharmacological sense, are  
66 usually named QSAR; more properly, they should be named QSPR (Katritzky et al., 1997) or QPPR (de  
67 Jongh et al., 1997), Quantitative Structure-Property and Quantitative Property-Property  
68 Relationships, respectively. For nanomaterials, such QSARs, QSPRs or QPPRs are still in the early  
69 stages of development, and are often named Quantitative Nanostructure-Activity Relationship  
70 (QNAR) or nano-QSPR; these include advanced statistical methods using machine learning (González-  
71 Durruthy et al., 2017). In this paper, we will use the term QPPR, which relates to all kinds of  
72 predictive relationships that use nanomaterial properties as descriptor. To date, a number of  
73 attempts have been made to correlate the characteristics of NMs to their biological responses

74 (Tantra et al., 2014; Raies and Bajic, 2016; Chen et al., 2017; Sizochenko and Leszczynski, 2017).  
75 Those reviews showed the tantalizing possibility that the QPPR method may indeed be feasible and  
76 useful in predicting the biological activity profiles of novel NMs. However, it also revealed that nano-  
77 QPPR is now still in its infancy and further challenges in this field need to be overcome. One issue  
78 standing out on this background relates to the comprehensive representation of NM structures. As  
79 known, NMs often exist as populations of materials varying in structural characteristics, e.g.  
80 composites, sizes, shapes, functional groups. The structural ambiguousness of NMs makes it difficult  
81 for experimentalists to provide precise information on NM characterization which consequently  
82 hinders the calculation of representative descriptors for NMs (Tamm et al., 2016).

83 Another issue of importance in this context concerns the dynamics of NMs in media. NMs often  
84 strongly interact with constituents in the medium and undergo dramatic changes to their surface  
85 properties, and dissolution and aggregation behavior (Winkler, 2016). These changes consequently  
86 alter the mobility, bioavailability, and ultimately the toxicity of NMs. Therefore, in some cases the  
87 toxicity information of NMs can be poorly correlated to the NMs' characteristics without considering  
88 the dynamics of NMs in the media. Thus QPPRs, predicting toxicity, based on initial structural  
89 features of NMs are now also extended to incorporate experimental descriptors like zeta-potential  
90 (Fourches et al., 2010; Liu et al., 2011; Singh and Gupta, 2014) and aggregate size (Sayes and Ivanov,  
91 2010; Sizochenko et al., 2014; Pan et al., 2016). The fact that these dynamics play a role in NM risk  
92 assessment was previously made clear in several studies focused on environmental exposure  
93 assessment (Westerhoff and Nowack, 2013; Cornelis, 2014; Hendren et al., 2015a; Baun et al., 2017).  
94 These studies suggest the use of empirical parameters for predicting risk, which include the effects of  
95 experimental conditions, such as pH, ionic strength and Natural Organic Matter content. Although  
96 several possibilities exist, an early study indicated that global descriptors for NM fate and transport  
97 need to include information on at least these experimental conditions (Westerhoff and Nowack,  
98 2013).

99 Just as for molecular or ionic chemical substances, other methods than QPPRs are available as well  
100 for predicting the safety of NMs. These alternatives include mechanistic models, tools which  
101 implement these models and overarching frameworks (Hristozov et al., 2016; Liguori et al., 2016;  
102 Sanchez Jimenez et al., 2016; Baun et al., 2017; Boyes et al., 2017; Nowack, 2017). These methods  
103 range from models based on commonly applied regulatory accepted approaches, predominantly in  
104 the area of exposure assessment, to more novel approaches such as used for hazard banding. The  
105 aim of all these tools, models and frameworks is to reduce the burden of testing NMs case by case  
106 and to focus on predicting risks based on the physico-chemical properties of a NM and on its  
107 application and use. The tools and models have NM properties as input parameters. Often the

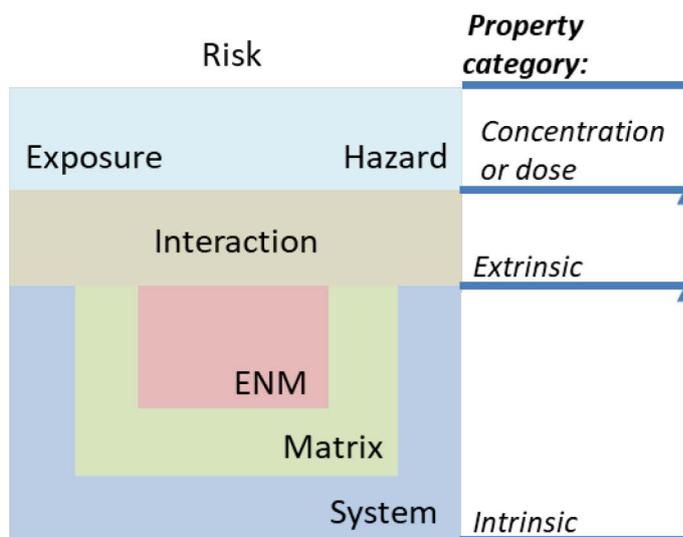
108 parametrization of the models is based on assumptions related to the processes that are deemed  
109 relevant based on the mechanistic understanding of the system and more pragmatic choices, e.g.  
110 based on data availability. This leads to differences but also to commonalities in processes and  
111 parameters used between the currently available models. In this respect the 'age' of a model is also  
112 an issue as it takes a considerable effort to keep them up to date with the most recent mechanistic  
113 understanding, with newer versions being developed almost continuously (Hristozov et al., 2016).

114 The key question with regard to the applicability of QPPRs in NM specific risk assessment is how  
115 QPPRs can be used to predict model parameters instead of requiring empirical data for each unique  
116 NM. To answer this question, we analyzed a selection of currently available mechanistic models and  
117 their parametrization related to nanomaterial properties. As such, we did not aim to include an  
118 exhaustive review of all available tools, models and methods for risk assessment, but we intended to  
119 include as many processes and parameters deemed relevant for risk assessment in order to provide a  
120 novel insight on the future of predictive risk assessment and the use of *in silico* methods in  
121 combination with mechanistic modelling.

122 The analyzed models predict hazard and/or external and internal exposure concentrations for  
123 humans and for the environment. For hazard assessment, only a few hazard banding tools could be  
124 analyzed as this is still almost solely based on (eco)toxicity testing. Although in hazard assessment  
125 other developments and discussions play an important role in reducing animal testing (e.g. *in vitro*-  
126 *in vivo* extrapolation), our analysis focuses on available hazard banding tools and *in silico* methods.

127 In the analysis of model parameters and QPPR descriptors a classification is introduced using three  
128 categories loosely based on the strategy for nanomaterial risk forecasting as presented by Hendren  
129 et al. (2015a), see Figure 1. The first category consists of the intrinsic properties of either the NM, the  
130 matrix or the (experimental) conditions of the system. The second, affected by the first, are the  
131 extrinsic NM properties that are based on the interaction of NMs with their surrounding matrix and  
132 the conditions affecting that system, e.g. a process rate constant or the attachment efficiency. The  
133 hazard concentrations (no effect, effect or lethal concentrations) or exposure concentrations  
134 (measured or otherwise estimated) make up the third category. These hazard and exposure  
135 concentrations are the regulatory basis for assessing risks. The division into these categories is used  
136 in the discussion and recommendations to assist in finding better descriptors for use in development  
137 of QPPRs and *in silico* methods with a basis in empirical data. For that reason the subcategorization  
138 of the intrinsic properties, relates to the fact that this metadata should be reported together with the  
139 main experimental outcome (Marchese Robinson et al., 2016). This is because the intrinsic NM,  
140 matrix and experimental system properties inherently affect the extrinsic NM properties and

141 eventually risk, in this sense the categorization is hierarchical (Figure 1). For example, an intrinsic NM  
142 property is size, an intrinsic matrix property is pH or ionic strength and an intrinsic experimental  
143 condition is temperature or mixing rate, these likely affect an extrinsic NM property such as the  
144 dissolution rate.



146

147 **Figure 1. Schematic representation of property categories and types of data required for predicting the risk of engineered**  
 148 **nanomaterials (NMs) based on quantitative relationships. These are (i) intrinsic properties of the ENM, the matrix and**  
 149 **the overall experimental System, (ii) Extrinsic properties that are dependent on the interaction between ENM, Matrix**  
 150 **and System conditions. (iii) Exposure and hazard concentrations used for risk assessment.**

## 151 2 Nanomaterial related properties in environmental exposure 152 models

153 Several environmental exposure models have been developed describing transport and  
 154 transformation processes of nanomaterials (Praetorius et al., 2012; Liu and Cohen, 2014; Meesters et  
 155 al., 2014; Dale et al., 2015a; Quik et al., 2015; Garner et al., 2017). We have analyzed the processes  
 156 reported in these fate models specifically affecting NM transport or transformation, to evaluate the  
 157 dependency of these models on intrinsic NM properties or extrinsic NM properties related to  
 158 interaction with the intrinsic matrix or system conditions (Table 1). Although numerous properties  
 159 affect NM fate, not all of them are related to intrinsic or extrinsic NM properties (Table S1 in  
 160 supporting information). For example, the description of dry and wet deposition and resuspension of  
 161 aerosols is mainly based on atmospheric characteristics such as the rain rate, wind speed and aerosol  
 162 properties (Nho-Kim, 2004; Wang et al., 2010). Runoff from soil to water and leaching out of the soil  
 163 are processes primarily related to soil characteristics and the rain rate (Renard et al., 1997). In the  
 164 aquatic compartment, sedimentation and resuspension can also be considered processes already  
 165 taken into account in multimedia modelling of conventional chemicals (Hollander et al., 2016). The  
 166 process rates are thought to be largely based on the characteristics of the sediment and natural  
 167 suspended particulate matter characteristics rather than on properties of the NM (Quik et al., 2012;  
 168 Dale et al., 2015b).

169 In practice, the effects that nanomaterials have on these processes cannot be fully neglected. It is  
170 clear from experimental and modelling studies that hetero-agglomeration between soil, sediment or  
171 suspended particles and NMs is an important process affecting their transport (Praetorius et al.,  
172 2012; El Badawy et al., 2013; Quik et al., 2014; Therezien et al., 2014; Bouchard et al., 2015; Quik et  
173 al., 2015; Ghosh et al., 2016). The mechanistic approach to including this process in models is to  
174 estimate the hetero-agglomeration rate, which is dependent on the hetero-agglomeration rate  
175 constant, which equals the product of the collision frequency and attachment efficiency (also called  
176 attachment affinity or attachment factor, and  $\alpha$ ) (Lyklema, 2005). There are theoretical approaches  
177 to calculate these properties (Lyklema, 2005; Petosa et al., 2010), but they mostly apply to relatively  
178 simple colloidal systems, e.g. not taking into account more complex behavior due to the presence of  
179 natural organic matter or protein corona's. For this reason the hetero-agglomeration rate or  
180 attachment efficiency is mostly measured empirically (Westerhoff and Nowack, 2013; Barton et al.,  
181 2014; Garner et al., 2017).

182 In addition to hetero-agglomeration affecting NM transport, the NMs are transformed into a new  
183 form, being the hetero-agglomerate. Several other transformations of NMs are deemed relevant in  
184 the natural environment, such as changes in the surface chemistry, disintegration due to chemical  
185 reactions and dissolution (Dale et al., 2015a). However, only dissolution and sometimes an overall  
186 process rate for additional degradation processes are included in the current models (Table 1).

187

188  
189  
190

**Table 1. The nanomaterial related parameters used for the reported parameterization methods commonly applied in nanomaterial fate models: Praetorius (Praetorius et al., 2012), MendNano (Liu and Cohen, 2014), SimpleBox4nano (Meesters et al., 2014), NanoDUFLOW (Quik et al., 2015) and NanoFate (Garner et al., 2017).**

Fate process	Model	Reported parametrization	Nanomaterial related properties
Dry deposition	MendNano	Theory for interception due to surface resistance combined with Stokes' Law	Size and density
	SimpleBox4nano	Theory for interception due to aerodynamic and surface resistance combined with Stokes' Law	Size and mass
	nanoFate	Stokes' law	Size and density
Wet deposition	MendNano	Below cloud rain scavenging ratio	No parameter, defined by size class
	SimpleBox4nano	Rain collection efficiency calculated from Brownian, Interception and gravitational impaction	Size and density
	nanoFate	Below cloud rain scavenging ratio	No parameter
Sedimentation	Praetorius, MendNano, SimpleBox4nano, NanoDUFLOW, nanoFate,	Stokes' Law	Size, density
Hetero-agglomeration (air)	MendNano	Fixed attachment and weighing factor	Attachment factor
	SimpleBox4nano	Coagulation coefficient and transitional correction coefficient or attachment efficiency	Size, density, attachment efficiency
	nanoFate	Empirically estimated hetero-agglomeration rate for freshwater adjusted based on lower collision frequency	Hetero-agglomeration rate constant
Hetero-agglomeration (water)	Praetorius, NanoDUFLOW, SimpleBox4nano	Smoluchowski theory on particle aggregation based on collision frequency and attachment efficiency	Size, density and attachment efficiency
	MendNano	Fixed attachment and weighing factor	Attachment factor
	nanoFate	Empirically estimated hetero-agglomeration rate	Hetero-agglomeration rate constant
Hetero-agglomeration (soil)	MendNano	Fixed attachment and weighing factor	Attachment factor
	SimpleBox4nano	Smoluchowski theory on particle aggregation and particle filtration theory with the attachment efficiency estimated empirically or using the interaction force boundary layer approximation	Size, density, attachment efficiency, surface potential, Hamaker constant
	nanoFate	Partitioning between solids and water fraction of soil based on empirical estimate of NM retention in soil.	NM-Soil retention fraction
Dissolution	MendNano	Based on solubility, mass transfer coefficient and available surface area of NMs	Concentration, size, density, fractal dimension
	NanoDUFLOW, SimpleBox4nano, nanoFate	Empirical	Dissolution rate constant
Agglomerate breakup	Praetorius, nanoDUFLOW,	No, assumed irreversible	No parameter

	MendNano, SimpleBox4nano, nanoFate		
Degradation and other transformation processes	Praetorius, MendNano, nanoFate	No	No parameter
	SimpleBox4nano, NanoDUFLOW	No	Degradation rate constant

191

192 From Table 1, it becomes clear that size and density are the only intrinsic NM properties used for  
 193 modelling the transport processes, deposition and sedimentation. The transformation processes do  
 194 not include other purely intrinsic nanomaterial related descriptors, except for the fractal dimension  
 195 of homo-aggregates in MendNano where homo-aggregates can be defined as the form of ENM under  
 196 consideration. The Hamaker constant, attachment efficiency, attachment factor, hetero-  
 197 agglomeration rate constant and dissolution rate constant are extrinsic properties not only related to  
 198 the nanomaterial, but also to the environmental compartment under consideration, including the  
 199 natural colloids and particulates.

### 200 3 Nanomaterial related properties in human exposure models

201 Human external exposure modelling traditionally largely depends on the application scenario of a  
 202 consumer product (in the case of consumer exposure) or on a worker's activity scenario (in case of  
 203 occupational exposure). For the former the calculation of the load of a NM based on the  
 204 concentration in a product and the frequency and amount of use are the relevant variables (RIVM,  
 205 2016) (see table S2 in supporting information), while for the latter the activity and the duration  
 206 mainly determine the exposure.

207 Although several models exist that take into account consumer exposure to chemicals and particles  
 208 due to inhalation from consumer products (sprays), only the well established Multiple-Path Particle  
 209 Dosimetry (MPPD) model and recent extension of ConsExpo nano  
 210 (<https://www.consexponano.nl/>)(RIVM, 2016) take specific nanomaterial properties into account  
 211 (Table 2). In both models the size and density, shape only in ConsExpo nano, of a NM are taken into  
 212 account for assessing the deposition of the NM in the lung. The NM dissolution rate is used to  
 213 estimate the clearance rate for soluble particles in ConsExpo-nano, for other particles both  
 214 ConsExpo-nano and MPPD use a particle independent clearance rate constant.

215 The main concern in relation to estimating human occupational exposure of NM is related to  
 216 inhalation (Schneider et al., 2011). For occupational exposure several risk control banding tools (e.g.  
 217 Stoffenmanager Nano, NanoSafer CB, Control Banding Nanotool) also include an estimate of the  
 218 exposure to nanomaterials (Liguori et al., 2016). The exposure estimate in these tools is largely based  
 219 on the application scenario and dustiness is the only NM related property used (Table 2). Dustiness is

220 measured using standard testing methods and is thought to be primarily related to the coating and  
 221 agglomeration of NM (Jensen et al., 2008; Schneider and Jensen, 2009).

222 This shows that although dustiness of a powder is related to intrinsic physico-chemical properties of  
 223 the NM, the attractive and repulsive forces affecting agglomeration are also dependent on the  
 224 systems conditions and matrix, e.g. moisture decreases dustiness and statically charged systems  
 225 increase dustiness, meaning that this is an extrinsic parameter describing an interaction (Jensen et  
 226 al., 2008; Schneider and Jensen, 2009; Koivisto et al., 2015; Levin et al., 2015).

227 **Table 2. Nanomaterial related properties used in estimating worker and consumer exposure to NM using a selection of**  
 228 **control banding tools (Zalk et al., 2009; Duuren-Schuurman et al., 2011; Jensen et al., 2014) and quantitative consumer**  
 229 **exposure models (Anjilvel and Asgharian, 1995; Asgharian and Price, 2007; RIVM, 2016).**

Process	Model	Reported parametrization	Nanomaterial related properties
Exposure at room level due to worker handling	Stoffen manager nano	Application scenario, dustiness, moisture	Dustiness
	NanoSafer CB	Application scenario, dustiness	Dustiness
	CB NanoTool	Application scenario, dustiness, mistiness	Dustiness
Inhalation of spray product	ConsExpo nano	Similar to (conventional) ConsExpo model	Not reported
	ConsExpo nano	ICRP deposition model	Size, density and shape
Deposition of NM in lungs	Multiple-Path Particle Dosimetry (MPPD)	Semi-emperical relationship using the molecular diffusion coefficient and effective diffusion coefficient in combination with lung dimensions	Size, density
	ConsExpo nano	ICRP clearance model for non-soluble particulates using clearance rate constants First order removal due to dissolution for soluble nano materials.	Dissolution rate constant
Clearance of NM from lungs	MPPD	Semi empirical relationship using clearance rate constants.	Not reported

230

## 231 4 Nanomaterial related properties in internal exposure/kinetic 232 models

233 Although modelling of internal concentrations of compounds has been applied in risk assessment of  
 234 chemicals, its use is often limited by availability of sufficiently generic data on the required input  
 235 parameters. As internalization of NMs is an important driver for NM toxicity, these types of models  
 236 are promising. Furthermore, this area of research contributes to future risk assessment methods that  
 237 depend less on *in vivo* studies. From data on the external exposure and intake of NMs, the internal  
 238 concentration in relevant organs in the human body can be calculated using physiologically based

239 pharmacokinetic (or PBPK) models (Lankveld et al. (2010), Bachler et al. (2013, 2014), Van Kesteren  
240 et al. (2015), Heringa et al. (2016), Li et al. (2017) and references cited herein (Lee et al., 2009; Péry  
241 et al., 2009; Li et al., 2012)). Currently, most PBPK models depend on NM specific parameters that  
242 were fitted from experimental data. The main process parameters are related to the absorption and  
243 distribution of NMs to different organs, the metabolism and the excretion of NMs (Table 3).

244 The absorption of NMs to the skin, lungs and intestines is commonly modelled using an absorption  
245 fraction that is fitted using experimental data (Table 3). So, no clear relationship with any intrinsic  
246 physico-chemical NM property is used, although it is expected that several intrinsic NM properties  
247 will affect the absorption fraction or rate, such as NM chemical composition and coating, but also the  
248 intrinsic characteristics of the biological surface will play a role. This is similar to how in  
249 environmental exposure modelling, the attachment affinity is based on both the NM and the other  
250 surface to which the NM will be attached (or in this case: absorbed).

251 The distribution of NMs via the blood to the different organs is based on organ uptake and release  
252 rates which are dependent on the formation of the protein corona, particle size, surface charge, and  
253 dissolution, speciation (Lankveld et al., 2010) and the crystalline form (van Kesteren et al., 2015;  
254 Heringa et al., 2016) of the NM (Table 3). However, one of the main processes governing this  
255 distribution is related to the ability of NMs to cross the capillary wall of the organs and by uptake by  
256 macrophages in the mononuclear phagocyte system (MPS) (Bachler et al., 2013, 2014). These  
257 macrophages are primarily located in the liver, lung and spleen. The former process of crossing the  
258 capillary wall was reported as size independent for the size range from 15 to 150 nm, whereas the  
259 latter (uptake by macrophages) is dependent on the size of the particle. The minor influence of size  
260 (for silver and TiO<sub>2</sub>)(Bachler et al., 2013, 2014), of the surface charge, of coating (for silver) (Bachler  
261 et al., 2013), and of the crystalline structure of the particles (TiO<sub>2</sub>)(Bachler et al., 2014) on the passing  
262 of the capillary wall of the organs may be explained by the formation of a protein corona. Thus, the  
263 extrinsic property of a protein corona may have a stronger influence on the distribution than the  
264 intrinsic NM properties (Bachler et al., 2013).

265 The metabolism of NMs is related to the dissolution of NMs (Table 3). For silver, the formation of  
266 silver sulfide particles was the main metabolic process. The formation of silver sulfide complexes  
267 caused storage of these particles in the different organs. For each organ the relative complexation  
268 capacity was estimated using the glutathione (GSH) content of the organs (Bachler et al., 2013).

269 The excretion of NMs is considered size independent, although different mechanisms are used for Ag  
270 or TiO<sub>2</sub> NMs (Bachler et al., 2013, 2014). The Ag NM excretion was due to the biliary endocytosis of  
271 silver-GSH complexes and for TiO<sub>2</sub> NMs this was due to the trans- capillary pathway.

272 In summary, estimating the internal concentration of nanomaterials is largely based on the physical  
 273 and biological characteristics of the bloodstream and different organs of the human body in  
 274 combination with NM characteristics. NM size and crystalline structure are found to be the only  
 275 intrinsic NM properties, and the other parameters are all extrinsic, related to the interaction of the  
 276 NM with the matrix and blood or organ system. In particular, the surface chemistry and the  
 277 formation of a protein corona could play an important role, e.g. in estimating the absorption to skin,  
 278 lungs and intestines. In environmental and colloid sciences, the attachment affinity is an important  
 279 similar property both related to the interaction of a NM and another surface with which the NM  
 280 interacts, e.g. sediment or soil particulates. It was also shown that the NM characteristics itself could  
 281 be of lesser importance compared to the interaction with proteins contained in the blood which  
 282 result in formation of a protein corona (Li et al., 2017). This means that these proteins should be  
 283 included in estimating any parameter related to absorption or attachment to biological surfaces.

284 **Table 3. Processes and nanomaterial related properties as reported in studies on pharmacokinetic models using**  
 285 **nanomaterials: Lankveld et al.(Lankveld et al., 2010), Bachler et al. (Bachler et al., 2013, 2014) and references cited**  
 286 **herein (Lee et al., 2009; Péry et al., 2009; Li et al., 2012).**

Process	Studied organ/compartment	Nanomaterial related properties
Absorption	Skin, lung, intestine	Absorption fraction to skin and intestine. Absorption rate to lung
Distribution	Blood and all organs	Size, surface charge, surface coating and protein corona, crystalline structure
Metabolism	Liver, lung, other organs	Dissolution rate, sulfidation rate
Excretion	Bile, kidney (urine), intestine	No reported dependencies

287

## 288 5 Nanomaterial related parameters in hazard banding tools

289 Although modelling is currently not common in estimating the hazard of chemicals in a regulatory  
 290 context, several control banding tools are available to perform a first risk screening of a NM  
 291 application in order to prioritize further assessment (Zalk et al., 2009; Duuren-Schuurman et al.,  
 292 2011; Höck J. et al., 2011; Jensen et al., 2014). The approaches of these authors to estimating NM  
 293 hazard are briefly, although not exhaustively, analyzed here. For a more thorough review see e.g.  
 294 Liguori et al. (2016), Sanchez Jimenez et al. (2016) or Hristozov et al. (2016).

295 In general, all the considered tools distinguish some parameter that indicates whether the NM is  
 296 expected to be persistent, often related to solubility. However, in the case of the Precautionary  
 297 Matrix, a more general classification based on a nanomaterial half-life is used. In this sense,  
 298 dissolution reflects the potential for degradation of NMs by an organism and not the potential

299 toxicity related to a transformation product, such as the dissolved ion. The second and most  
 300 important aspect in estimating NM hazard is classifying the potential toxicity of a NM based on either  
 301 physico-chemical characteristics and/or the toxicological characteristics of the NM (toxic potential,  
 302 Table 4). Several tools also include the toxicological characteristics of either bulk, larger sized  
 303 particulates or of the parent chemical compound for estimating the toxic potential. Although there is  
 304 some variation between the intrinsic physico-chemical characteristics considered, 3 out of 4 tools use  
 305 shape to classify the toxic potential of a NM. They consider fibrous or tubular particles with a high  
 306 aspect ratio to coincide with a high toxic potential following the fibre paradigm (Poland et al., 2008).  
 307 Furthermore, 3 out of 4 tools include NM surface chemistry, for which the parametrization ranges  
 308 from classifying the catalytic or redox potential to identifying the presence of surface  
 309 coatings/modifications. Only one of the tools considers size a driver for toxic potential. For the  
 310 eventual risk assessment, NanoSafer CB and ConsExpo use the specific surface area, based on NM  
 311 size and density, to scale the exposure limits and exposure potentials.

312 This analysis shows that empirical toxicity data are a main component of hazard assessment, also in  
 313 these hazard banding tools (hazard concentrations in Figure 1). Although it is clear that several  
 314 inherent relationships between adverse effects and intrinsic NM physico-chemical properties are  
 315 taken into account, only a few parameters relate to the interaction between a NM and the matrix or  
 316 system, e.g. solubility or half-life.

317 **Table 4. Nanomaterial related properties included in processes affecting hazard estimation used in risk control banding**  
 318 **tools.**

Process	Model	Reported classification method	Nanomaterial related properties
Biopersistence	Stoffenmanager nano	Soluble/insoluble	Solubility
	NanoSafer CB	Soluble/insoluble	Solubility
	Precautionary Matrix	NM stability	Half-life
	CB NanoTool	Soluble/insoluble	Solubility
Toxic potential	NanoSafer CB	Occupational exposure limit or risk sentence of conventional analogue compound, shape and coating	Dimension of primary particle, presence of coating/surface modification, hazard data
	Stoffen manager nano	Fiber aspect ratio, Hazard band NM and/or parent material based on hazard classification for either carcinogenicity, mutagenicity, reproduction toxicity or sensitisation.	Fiber aspect ratio, hazard data
	Precautionary Matrix	Classification of catalytic & redox activity	Catalytic & redox activity
	CB NanoTool	Mutagenicity, carcinogenicity, dermal, and reproductive effects of parent and micron-size	Surface activity, solubility, shape, size, hazard data

		or NM, surface activity, particle shape, particle diameter	
Hazard concentration	n-SSWD	Ecotoxicological data corrected for Species relevance, trophic level abundance and data quality	Hazard data

319

## 320 6 Discussion

### 321 6.1 Role of QPPRs in risk assessment

322 Based on the models and assessment methods currently used for prognostic risk assessment as  
323 presented here, it should be clear that QPPRs have different goals in exposure assessment on the  
324 one hand and hazard assessment on the other. Whereas exposure assessment uses quantitative  
325 mechanistic modelling techniques, hazard assessment mainly depends on measuring toxicity or using  
326 tools for a more qualitative hazard assessment. This means that, for hazard assessment, *in silico*  
327 methods, such as QPPRs, are most useful for predicting hazard concentrations (see supporting info  
328 table S4). On the other hand, for the exposure models - based on intrinsic and extrinsic NM  
329 properties - QPPRs and *in silico* methods in general are valuable for the estimation of extrinsic input  
330 parameters. Several of such hazard QPPRs have been developed, either classifying NMs into hazard  
331 categories or quantitatively predicting toxicity, which provide useful output for hazard band tools  
332 and hazard assessment in general. Only a few exposure-related nano-QPPRs have been developed,  
333 such as those for predicting the zeta-potential (Mikolajczyk et al., 2015; Wyrzykowska et al., 2016).  
334 Most other currently available QPPRs predict parameters that are not used for risk assessment of  
335 NMs, e.g. related to adsorption of compounds to NMs (Heidari and Fatemi, 2016; Toropova and  
336 Toropov, 2016; Urbaszek et al., 2017) or the  $K_{ow}$  of carbon nanotubes (Toropov et al., 2007). Here lies  
337 an opportunity to develop new *in silico* methods to predict different interactions of NM and natural  
338 and biological surfaces, such as the attachment efficiency, whereas it is more logical to use a  
339 modelling approach when the full mechanistic functioning of a system is understood. The strength of  
340 using nano-QPPRs here lies in bridging the gap between a NM property and a model parameter when  
341 this relationship is not (easily) quantifiable. This is for example the case for predicting the attachment  
342 efficiency. As mentioned above, the attachment efficiency can be calculated based on measurements  
343 of the zeta-potential, the Hamaker constant and NM radius (Petosa et al., 2010; Meesters, 2017).  
344 However, this is only valid for ideal systems where complexities due to the presence of proteins or  
345 natural organic matter and variable properties of natural and biological surfaces do not play a role  
346 (Petosa et al., 2010). This makes mechanistic modelling less relevant for environmental or biological  
347 systems. For this reason, development of a QPPR predicting the attachment efficiency, based on data  
348 gathered using a range of empirical data, appears a more beneficial approach. In addition, it is to be  
349 noted that use of empirical data on other transformation processes, such as dissolution, should be  
350 considered for QPPRs and other types of *in silico* methods, such as material modelling. Material  
351 modelling, for example, has been used for predicting dissolution kinetics of active pharmaceutical  
352 ingredients (Elts et al., 2016).

353 A major drawback in current efforts of developing any *in silico* method based on empirical data, is  
354 the present low availability and quality of data. For this reason, current activities have shown very  
355 limited success due to data scarcity, non-standardized testing methods and incomplete reporting of  
356 the NM, of the matrix used, and of experimental conditions (Hendren et al., 2015b; Marchese  
357 Robinson et al., 2016). Such development will only work when standardized assays are used in order  
358 to combine datasets for QPPR development and thus to allow for optimal use of data from different  
359 studies. Additionally, data curation systems need to be used, such as those developed for the  
360 Nanomaterials Registry (Guzan et al., 2013) and the NanoInformatics Knowledge Commons  
361 (<https://ceint.duke.edu/research/nikc>).

362 Based on the current understanding of NM behavior in the environment and in humans and  
363 organisms, it should be clear that their interaction with the surroundings is an important aspect to  
364 consider. Extrinsic parameters are the drivers of most exposure models (see Table 1, 2 and 3). Only  
365 few of these extrinsic parameters such as the sedimentation velocity can be estimated based on a  
366 quantitative theory using solely intrinsic parameters. These intrinsic parameters (such as size, shape  
367 and density) reported in earlier studies on pristine particles can be used for mechanistic modelling,  
368 although, as stated earlier, curation of reported data needs attention (Hendren et al., 2015b). Most  
369 parameters need to be estimated empirically in the relevant systems (Westerhoff and Nowack, 2013;  
370 Hendren et al., 2015a; Geitner et al., 2016). These extrinsic parameters which describe the  
371 interaction between the nanomaterial, the matrix in which they are present and the system's  
372 conditions (Figure 1) are inherently dependent on more than the properties of the nanomaterial  
373 alone. This means that any QPPR aimed at predicting extrinsic parameters should include intrinsic  
374 descriptors related to the matrix and system characteristics. This can be done using 'easily' measured  
375 extrinsic descriptors, e.g. zeta-potential or aggregate size, or by using intrinsic descriptors that also  
376 include system and matrix characteristics.

377 In the analysis of the environmental exposure models we have identified the most important  
378 extrinsic input parameters to be the attachment efficiency or the hetero-agglomeration rate  
379 constant, the dissolution rate constant, and rate constants related to transformation or degradation.  
380 In human exposure models dustiness is a key parameter, whereas also the absorption rate, surface  
381 charge and coating affect the internal exposure concentration (Table 5). These properties can all be  
382 empirically estimated, but this often requires a significant monetary investment. For this reason,  
383 further mechanistic understanding is needed on the interactions between NMs, the matrix and  
384 systems conditions, in order to find easily measured descriptors or parameters that can function as a  
385 basis for estimating these input parameters. This is the focus of the current efforts to develop  
386 specific standardized assays for these relationships between an input parameter and readily available

387 characteristics of the experimental system, commonly called functional assays (Hendren et al.,  
388 2015a). In addition to using such standardized empirical methods or functional assays, the resulting  
389 data should be available so that they can be used to create *in silico* methods such as QPPRs.  
390 Eventually this will lead to a link between the extrinsic process parameters or interaction descriptors  
391 and intrinsic descriptors based on the NM, matrix and system. In this way, input parameters used for  
392 mechanistic exposure modelling can be estimated without the need for further functional assays.  
393 These input parameters can rather be estimated using much simpler measurements of specific  
394 properties, similar to how  $K_{ow}$  is used for organic compounds. This relationship between different  
395 types of descriptors and relevant input parameters is important to realize in further development of  
396 nano-QPPRs and other *in silico* models.

## 397 6.2 Descriptors in hazard assessment

398 From analyzing the current state-of-the-art of QPPRs for metal-based NM as reviewed in Chen *et al.*  
399 (2017), it is clear that hazards of NMs are dependent on a variety of intrinsic NM properties, some  
400 experimental conditions such as NM concentration, and on extrinsic characteristics of the interaction  
401 of the NM with the target organism or exposure matrix. These extrinsic characteristics include zeta-  
402 potential, agglomerate size in the exposure media or water and in one case the overlap of  
403 conduction band energy levels with the cellular redox potential and solubility. It is clear that these  
404 types of characteristics of the interaction of NM with organisms are important and the search for  
405 better descriptors (Tamm et al., 2016; Toropova et al., 2016) should include them in addition to  
406 intrinsic NM descriptors (Chen et al., 2017; Shityakov et al., 2017).

407 Given the analysis of the parametrization of hazard assessment models, we conclude that mainly  
408 hazard data itself, based on dose response relationships are important in assessing the toxic  
409 potential of NMs (Table 4). Furthermore, the intrinsic NM properties such as fiber aspect ratio, size,  
410 coating, and surface activity are used as drivers for this toxic potential. In only one tool, solubility is  
411 included in relation to the toxic potential, and all the tools otherwise use solubility or half-life only to  
412 estimate the persistence. In most cases this means that NM with high solubility would not be  
413 considered according to the NM specific toxic potential estimate, but related to the conventional  
414 compound.

415 In comparison with the hazard band tools, none of the QPPRs takes the fiber aspect ratio or shape  
416 into account. Although QPPRs are developed for high aspect ratio NMs, such as carbon nanotubes,  
417 they are only applicable to these types of NM and also do not include size or aspect ratio as a  
418 descriptor (González-Durruthy et al., 2017). Overall, both the hazard band tools and nano-QPPRs  
419 have descriptors related to the surface activity of NMs. Most currently available QPPRs have rather

420 narrow applicability domains, e.g. limited to one core material with different coatings or different  
421 cores, but similar shape and coating, see table S3 in supporting info. Using a broader set of  
422 descriptors based on the known NM toxicity mechanisms could extend this applicability domain.

423 In addition to the different NM related properties that affect the toxic potential, the eventual  
424 adverse effects are also related to the kinetics of the uptake and internal distribution processes of  
425 the NM in humans and organisms. This means that any important parameter or descriptor identified  
426 in those studies (Table 3) is likely to also drive the hazard of an NM. This shows that three important  
427 interactions likely play a role in hazard assessment, but have until now not been commonly  
428 parameterized in hazard banding tools or as descriptors in QPPRs. The first of these is the interaction  
429 of NMs with organs, such as absorption to skin and lungs affecting internalization. Empirically  
430 estimating the attachment efficiency could prove useful here. The biological relevance of this  
431 parameter has recently been shown in a study on the trophic transfer of NMs through the food chain  
432 of aquatic organisms (Geitner et al., 2016). The second interaction of importance is the formation of  
433 a protein corona affecting several processes related to the distribution of NMs between blood and  
434 organs. This interaction is related to the first key interaction identified, but the focus here is on the  
435 formation and stability surrounding a protein corona and the NM itself. Formation of a protein  
436 corona affects the attachment efficiency, but other descriptors are likely to be relevant in this  
437 respect as well. The third interaction is the degradation of NM to other forms, e.g. dissolved ionic  
438 species (Waalewijn-Kool et al., 2013; Schwabe et al., 2014) or metabolites (Levard et al., 2011; Hou et  
439 al., 2015).

### 440 6.3 Conclusion

441 In conclusion, there is a big difference in the models and tools available to predict exposure or hazard  
442 of ENMs. This is mainly due to the more qualitative approach commonly applied to predicting hazard  
443 compared to the quantitative estimates of exposure. However, the currently used set of parameters  
444 for both exposure and hazard assessment is limited in nature, and consists of intrinsic and extrinsic  
445 parameters related to the dynamic interactions between NMs and the exposure media or biological  
446 kinetics (Table 5). These often complex interaction processes related to hazard or exposure can  
447 inherently be described using descriptors for the intrinsic characteristics of the NM, matrix and  
448 system conditions or by simpler extrinsic descriptors of interaction. This could for example be  
449 relationships between the aggregation rate and pH, organic matter concentration and ionic strength  
450 (Liu et al., 2013) or between the zeta-potential, an easily measured interaction type parameter, and  
451 the attachment efficiency (Wang and Keller, 2009). These relationships should be quantifiable using  
452 *in silico* methods, such as QPPRs and other modeling approaches, based on empirical datasets from  
453 standardized functional assays. This also means that the required data should be made available for

454 the *in silico* modeling research field. These data should consist not only of the measured parameter,  
 455 such as NM size, attachment efficiency or hazard concentration, but include meta-data that covers  
 456 the relevant intrinsic properties of the NM, matrix and experimental conditions (Figure 1).

457 **Table 5. Overview of nanomaterial related model parameters used in the analyzed models to predict nanomaterial risk.**  
 458 **Italic parameters are likely to be useful endpoints for QPPRs and underlined parameters are likely descriptors for QPPRs.**  
 459 **This should not be seen as a limitative list, specifically for the ENM intrinsic properties.**

<b>Hazard concentration</b>	<b>NM extrinsic property</b>	<b>NM intrinsic property</b>
<i>ECx, LCx, NOEC</i>	<u><i>Attachment efficiency</i></u>	<u>Size</u>
	<u><i>NM-Soil retention factor</i></u>	<u>Density</u>
	<u><i>Hetero-agglomeration rate</i></u>	<u>Shape</u>
	<u><i>Absorption rate</i></u>	<u>Coating</u>
	<u>Dustiness</u>	
		<u>Surface activity</u>
	<u><i>Dissolution rate</i></u>	<u>Catalytic &amp; redox</u>
	<u><i>Sulfidation rate</i></u>	<u>activity</u>
	<u><i>Degradation rate</i></u>	<u>Isoelectric Point</u>
		<u>Crystalline structure</u>
	<u>Zeta potential</u>	
	<u>Surface Charge</u>	
	<u>Hamaker constant</u>	

460

461 Given the inherent relationship between NM properties and the interaction with the relevant matrix  
 462 or organism it can be hypothesized that even though changes of NM properties could occur in the  
 463 exposure media, the characteristics of the pristine NMs may still be linked to the observed adverse  
 464 biological effects or transformation and behavior. However, the current understanding of these  
 465 complex interactions requires the use of descriptors related to the interaction of NM and the  
 466 relevant exposure matrix. Although several descriptors are identified here based on the parameters  
 467 used in modelling (Table 5), further steps are needed in finding relevant descriptors and developing  
 468 better QPPRs in general. These steps include (i) availability of standardized methods for measuring  
 469 the interaction parameters. The methods need to include proper characterization of NM properties  
 470 and proper reporting of the matrix and of the experimental conditions; (ii) improvement of the  
 471 availability of new and existing data for modeling, e.g. using the current state-of-the-art data systems  
 472 including data curation to improve data quality (Thomas et al., 2013; Hastings et al., 2015; Hendren  
 473 et al., 2015b). Overall, this should lead to novel risk assessment tools, which incorporate improved *in*  
 474 *silico* models. These novel tools should be validated with high quality data so they can be accepted  
 475 for regulatory use.

## 476 7 Acknowledgements

477 We thank Keld Alstrup Jensen, Walter Brand, Ilse Goosens, Minne Heringa, and Dick de Zwart for  
478 their contribution as well as numerous other colleagues for the many fruitful discussions. This work is  
479 supported by funding from the European Union's Horizon 2020 research and innovation programme  
480 under grant agreement No 686239 "caLIBRAte" and by NanoNextNL, a micro- and nanotechnology  
481 consortium of the Government of The Netherlands and 130 partners.

## 482 8 References

- 483 Anjilvel, S., Asgharian, B., 1995. A Multiple-Path Model of Particle Deposition in the Rat Lung.  
484 *Fundamental and Applied Toxicology* 28, 41-50.
- 485 Asgharian, B., Price, O.T., 2007. Deposition of ultrafine (nano) particles in the human lung. *Inhal*  
486 *Toxicol* 19, 1045-1054.
- 487 Bachler, G., von Goetz, N., Hungerbuhler, K., 2013. A physiologically based pharmacokinetic model  
488 for ionic silver and silver nanoparticles. *Int J Nanomedicine* 8, 3365-3382.
- 489 Bachler, G., von Goetz, N., Hungerbuhler, K., 2014. Using physiologically based pharmacokinetic  
490 (PBPK) modeling for dietary risk assessment of titanium dioxide (TiO) nanoparticles. *Nanotoxicology*,  
491 1-8.
- 492 Barton, L.E., Therezien, M., Auffan, M., Bottero, J.-Y., Wiesner, M.R., 2014. Theory and Methodology  
493 for Determining Nanoparticle Affinity for Heteroaggregation in Environmental Matrices Using Batch  
494 Measurements. *Environmental Engineering Science* 31, 421-427.
- 495 Baun, A., Sayre, P., Steinhäuser, K.G., Rose, J., 2017. Regulatory relevant and reliable methods and  
496 data for determining the environmental fate of manufactured nanomaterials. *NanoImpact* 8, 1-10.
- 497 Bouchard, D., Chang, X., Chowdhury, I., 2015. Heteroaggregation of multiwalled carbon nanotubes  
498 with sediments. *Environmental Nanotechnology, Monitoring & Management*.
- 499 Boyes, W.K., Thornton, B.L.M., Al-Abed, S.R., Andersen, C.P., Bouchard, D.C., Burgess, R.M., Hubal,  
500 E.A.C., Ho, K.T., Hughes, M.F., Kitchin, K., Reichman, J.R., Rogers, K.R., Ross, J.A., Rygielwicz, P.T.,  
501 Scheckel, K.G., Thai, S.F., Zepp, R.G., Zucker, R.M., 2017. A comprehensive framework for evaluating  
502 the environmental health and safety implications of engineered nanomaterials. *Crit Rev Toxicol*, 1-44.
- 503 Chen, G., Peijnenburg, W., Xiao, Y., Vijver, M.G., 2017. Current Knowledge on the Use of  
504 Computational Toxicology in Hazard Assessment of Metallic Engineered Nanomaterials. *Int J Mol Sci*  
505 18.
- 506 Cornelis, G., 2014. Fate descriptors for engineered nanoparticles: the good, the bad, and the ugly.  
507 *Environ. Sci.: Nano*.
- 508 Dale, A.L., Casman, E.A., Lowry, G.V., Lead, J.R., Viparelli, E., Baalousha, M., 2015a. Modeling  
509 nanomaterial environmental fate in aquatic systems. *Environ Sci Technol* 49, 2587-2593.
- 510 Dale, A.L., Lowry, G.V., Casman, E.A., 2015b. Stream Dynamics and Chemical Transformations Control  
511 the Environmental Fate of Silver and Zinc Oxide Nanoparticles in a Watershed-Scale Model. *Environ*  
512 *Sci Technol*.
- 513 de Jongh, J., Verhaar, H.J., Hermens, J.L., 1997. A quantitative property-property relationship (QPPR)  
514 approach to estimate in vitro tissue-blood partition coefficients of organic chemicals in rats and  
515 humans. *Archives of Toxicology* 72, 17-25.
- 516 Dearden, J.C., 2003. In silico prediction of drug toxicity. *J Comput Aided Mol Des* 17, 119-127.
- 517 Duuren-Schuurman, B., Vink, S., Brouwer, D., Kroese, D., Heussen, H., Verbist, K., Tielemans, E.,  
518 Niftrik, M.v., Fransman, W., 2011. *Stoffenmanager nano: Description of the conceptual control*  
519 *banding model*. TNO, Zeist, The Netherlands.
- 520 ECHA, 2008. Guidance on information requirements and chemical safety assessment - Chapter R.6:  
521 QSARs and grouping of chemicals. European Chemicals Agency, Helsinki, Finland, p. 134.

522 El Badawy, A.M., Hassan, A.A., Scheckel, K.G., Suidan, M.T., Tolaymat, T.M., 2013. Key factors  
523 controlling the transport of silver nanoparticles in porous media. *Environ Sci Technol* 47, 4039-4045.

524 Elts, E., Greiner, M., Briesen, H., 2016. Predicting Dissolution Kinetics for Active Pharmaceutical  
525 Ingredients on the Basis of Their Molecular Structures. *Crystal Growth & Design* 16, 4154-4164.

526 European Commission, 2006. Regulation (EC) No 1907/2006 of 18 December 2006 concerning the  
527 Registration, Evaluation, Authorisation and Restriction of Chemicals (REACH), establishing a European  
528 Chemicals Agency, amending Directive 1999/45/EC and repealing Council Regulation (EEC) No 793/93  
529 and Commission Regulation (EC) No 1488/94 as well as Council Directive 76/769/EEC and  
530 Commission Directives 91/155/EEC, 93/67/EEC, 93/105/EC and 2000/21/EC. in: *European*  
531 *Commission (Ed.). 2006R1907— EN— 12.10.2008 — 002.001*, Brussels, Belgium, p. 362.

532 Fourches, D., Pu, D., Tassa, C., Weissleder, R., Shaw, S.Y., Mumper, R.J., Tropsha, A., 2010.  
533 Quantitative nanostructure-activity relationship modeling. *ACS nano* 4, 5703-5712.

534 Fujita, T., Winkler, D.A., 2016. Understanding the Roles of the “Two QSARs”. *Journal of Chemical*  
535 *Information and Modeling* 56, 269-274.

536 Garner, K.L., Suh, S., Keller, A.A., 2017. Assessing the Risk of Engineered Nanomaterials in the  
537 Environment: development and application of the nanoFate model. *Environ Sci Technol*.

538 Geitner, N.K., Marinakos, S.M., Guo, C., O'Brien, N., Wiesner, M.R., 2016. Nanoparticle Surface  
539 Affinity as a Predictor of Trophic Transfer. *Environ Sci Technol* 50, 6663-6669.

540 Ghosh, S., Pradhan, N.R., Mashayekhi, H., Zhang, Q., Pan, B., Xing, B., 2016. Colloidal aggregation and  
541 structural assembly of aspect ratio variant goethite ( $\alpha$ -FeOOH) with nC60 fullerene in  
542 environmental media. *Environmental pollution*.

543 González-Durruthy, M., Alberici, L.C., Curti, C., Naal, Z., Atique-Sawazaki, D.T., Vázquez-Naya, J.M.,  
544 González-Díaz, H., Munteanu, C.R., 2017. Experimental–Computational Study of Carbon Nanotube  
545 Effects on Mitochondrial Respiration: In Silico Nano-QSPR Machine Learning Models Based on New  
546 Raman Spectra Transform with Markov–Shannon Entropy Invariants. *Journal of Chemical Information*  
547 *and Modeling* 57, 1029-1044.

548 Guzan, K.A., Mills, K.C., Gupta, V., Murry, D., Scheier, C.N., Willis, D.A., Ostraat, M.L., 2013.  
549 Integration of data: the Nanomaterial Registry project and data curation. *Computational Science &*  
550 *Discovery* 6, 014007.

551 Hastings, J., Jeliaskova, N., Owen, G., Tsiliki, G., Munteanu, C.R., Steinbeck, C., Willighagen, E., 2015.  
552 eNanoMapper: harnessing ontologies to enable data integration for nanomaterial risk assessment. *J*  
553 *Biomed Semantics* 6, 10.

554 Heidari, A., Fatemi, M.H., 2016. Hybrid Docking-Nano-QSPR: An Alternative Approach for Prediction  
555 of Chemicals Adsorption on Nanoparticles. *Nano* 11, 1650078.

556 Hendren, C.O., Lowry, G.V., Unrine, J.M., Wiesner, M.R., 2015a. A functional assay-based strategy for  
557 nanomaterial risk forecasting. *The Science of the total environment* 536, 1029-1037.

558 Hendren, C.O., Powers, C.M., Hoover, M.D., Harper, S.L., 2015b. The Nanomaterial Data Curation  
559 Initiative: A collaborative approach to assessing, evaluating, and advancing the state of the field.  
560 *Beilstein journal of nanotechnology* 6, 1752-1762.

561 Heringa, M.B., Geraets, L., van Eijkeren, J.C., Vandebriel, R.J., de Jong, W.H., Oomen, A.G., 2016. Risk  
562 assessment of titanium dioxide nanoparticles via oral exposure, including toxicokinetic  
563 considerations. *Nanotoxicology* 10, 1515-1525.

564 Höck J., Epprecht T., Furrer E., Hofmann H., Höhner K., Krug H., Lorenz C., Limbach L., Gehr P.,  
565 Nowack B., Riediker M., Schirmer K., Schmid B., Som C., Stark W., Studer C., Ulrich A., von Götz N.,  
566 Weber A., Wengert S., Wick P., 2011. Guidelines on the Precautionary Matrix for Synthetic  
567 Nanomaterials. Swiss Federal Office of Public Health and Federal Office for the Environment, Berne.

568 Hollander, A., Schoorl, M., van de Meent, D., 2016. SimpleBox 4.0: Improving the model while  
569 keeping it simple. *Chemosphere* 148, 99-107.

570 Hou, W.C., Chowdhury, I., Goodwin, D.G., Jr., Henderson, W.M., Fairbrother, D.H., Bouchard, D.,  
571 Zepp, R.G., 2015. Photochemical transformation of graphene oxide in sunlight. *Environ Sci Technol*  
572 49, 3435-3443.

573 Hristozov, D., Gottardo, S., Semenzin, E., Oomen, A., Bos, P., Peijnenburg, W., van Tongeren, M.,  
574 Nowack, B., Hunt, N., Brunelli, A., Scott-Fordsmand, J.J., Tran, L., Marcomini, A., 2016. Frameworks  
575 and tools for risk assessment of manufactured nanomaterials. *Environ Int* 95, 36-53.

576 Jensen, K.A., Koponen, I.K., Clausen, P.A., Schneider, T., 2008. Dustiness behaviour of loose and  
577 compacted Bentonite and organoclay powders: What is the difference in exposure risk? *Journal of*  
578 *Nanoparticle Research* 11, 133-146.

579 Jensen, K.A., Saber, A.T., Kristensen, H.V., Liguori, B., Jensen, A.C.Ø., Koponen, I.K., Wallin, H., 2014.  
580 NanoSafer vs. 1.1 Nanomaterial risk assessment using first order modeling. Poster session presented  
581 at Topical Scientific Workshop on Regulatory Challenges in Risk Assessment of Nanomaterials,  
582 Helsinki, Finland.

583 Katritzky, A.R., Karelson, M., Lobanov, V.S., 1997. QSPR as a means of predicting and understanding  
584 chemical and physical properties in terms of structure

585 Alan R. Katritzky, Mati Karelson and Victor S. Lobanova. *Pure & Appl. Chem.* 69, 245-248.

586 Koivisto, A.J., Jensen, A.C., Levin, M., Kling, K.I., Maso, M.D., Nielsen, S.H., Jensen, K.A., Koponen, I.K.,  
587 2015. Testing the near field/far field model performance for prediction of particulate matter  
588 emissions in a paint factory. *Environ Sci Process Impacts* 17, 62-73.

589 Lankveld, D.P.K., Oomen, A.G., Krystek, P., Neigh, A., Troost – de Jong, A., Noorlander, C.W., Van  
590 Eijkeren, J.C.H., Geertsma, R.E., De Jong, W.H., 2010. The kinetics of the tissue distribution of silver  
591 nanoparticles of different sizes. *Biomaterials* 31, 8350-8361.

592 Lee, H.A., Leavens, T.L., Mason, S.E., Monteiro-Riviere, N.A., Riviere, J.E., 2009. Comparison of  
593 Quantum Dot Biodistribution with a Blood-Flow-Limited Physiologically Based Pharmacokinetic  
594 Model. *Nano letters* 9, 794-799.

595 Levard, C.m., Reinsch, B.C., Michel, F.M., Oumahi, C., Lowry, G.V., Brown, G.E., 2011. Sulfidation  
596 Processes of PVP-Coated Silver Nanoparticles in Aqueous Solution: Impact on Dissolution Rate.  
597 *Environmental Science & Technology* 45, 5260-5266.

598 Levin, M., Rojas, E., Vanhala, E., Vippola, M., Liguori, B., Kling, K.I., Koponen, I.K., Møhlhave, K., Tuomi,  
599 T., Gregurec, D., Moya, S., Jensen, K.A., 2015. Influence of relative humidity and physical load during  
600 storage on dustiness of inorganic nanomaterials: implications for testing and risk assessment. *Journal*  
601 *of Nanoparticle Research* 17.

602 Li, M., Panagi, Z., Avgoustakis, K., Reineke, J., 2012. Physiologically based pharmacokinetic modeling  
603 of PLGA nanoparticles with varied mPEG content. *Int J Nanomedicine* 7, 1345-1356.

604 Li, M., Zou, P., Tyner, K., Lee, S., 2017. Physiologically Based Pharmacokinetic (PBPK) Modeling of  
605 Pharmaceutical Nanoparticles. *AAPS J* 19, 26-42.

606 Liguori, B., Hansen, S.F., Baun, A., Jensen, K.A., 2016. Control banding tools for occupational exposure  
607 assessment of nanomaterials — Ready for use in a regulatory context? *NanoImpact* 2, 1-17.

608 Liu, H.H., Cohen, Y., 2014. Multimedia Environmental Distribution of Engineered Nanomaterials.  
609 *Environmental Science & Technology* 48, 3281-3292.

610 Liu, J., Legros, S., von der Kammer, F., Hofmann, T., 2013. Natural organic matter concentration and  
611 hydrochemistry influence aggregation kinetics of functionalized engineered nanoparticles. *Environ*  
612 *Sci Technol* 47, 4113-4120.

613 Liu, R., Rallo, R., George, S., Ji, Z., Nair, S., Nel, A.E., Cohen, Y., 2011. Classification NanoSAR  
614 development for cytotoxicity of metal oxide nanoparticles. *Small* 7, 1118-1126.

615 Lyklema, J., 2005. Pair Interactions. in: Lyklema, J. (Ed.). *Fundamentals of Interface and Colloid*  
616 *Science, Volume IV, Particulate Colloids.* Elsevier Academic Press, Amsterdam.

617 Marchese Robinson, R.L., Lynch, I., Peijnenburg, W., Rumble, J., Klaessig, F., Marquardt, C., Rauscher,  
618 H., Puzyn, T., Purian, R., Aberg, C., Karcher, S., Vriens, H., Hoet, P., Hoover, M.D., Hendren, C.O.,  
619 Harper, S.L., 2016. How should the completeness and quality of curated nanomaterial data be  
620 evaluated? *Nanoscale* 8, 9919-9943.

621 Meesters, J.A.J., 2017. *Environmental Exposure Modeling of Nanoparticles.* Environmental Science.  
622 Radboud University Nijmegen, Nijmegen, p. 286.

623 Meesters, J.A.J., Koelmans, A.A., Quik, J.T.K., Hendriks, A.J., van de Meent, D., 2014. Multimedia  
624 Modeling of Engineered Nanoparticles with SimpleBox4nano: Model Definition and Evaluation.  
625 Environmental Science & Technology 48, 5726-5736.

626 Mikolajczyk, A., Gajewicz, A., Rasulev, B., Schaeublin, N., Maurer-Gardner, E., Hussain, S., Leszczynski,  
627 J., Puzyn, T., 2015. Zeta Potential for Metal Oxide Nanoparticles: A Predictive Model Developed by a  
628 Nano-Quantitative Structure–Property Relationship Approach. Chemistry of Materials 27, 2400-2407.

629 Nho-Kim, E., 2004. Parameterization of size-dependent particle dry deposition velocities for global  
630 modeling. Atmospheric Environment 38, 1933-1942.

631 Nowack, B., 2017. Evaluation of environmental exposure models for engineered nanomaterials in a  
632 regulatory context. Nanolmpact.

633 Pan, Y., Li, T., Cheng, J., Telesca, D., Zink, J.I., Jiang, J., 2016. Nano-QSAR modeling for predicting the  
634 cytotoxicity of metal oxide nanoparticles using novel descriptors. RSC Adv. 6, 25766-25775.

635 Péry, A.R.R., Brochot, C., Hoet, P.H.M., Nemmar, A., Bois, F.Y., 2009. Development of a physiologically  
636 based kinetic model for 99m-Tc-labelled carbon nanoparticles inhaled by humans.  
637 Inhalation Toxicology 21, 1099-1107.

638 Petosa, A.R., Jaisi, D.P., Quevedo, I.R., Elimelech, M., Tufenkji, N., 2010. Aggregation and Deposition  
639 of Engineered Nanomaterials in Aquatic Environments: Role of Physicochemical Interactions.  
640 Environmental Science & Technology 44, 6532-6549.

641 Poland, C.A., Duffin, R., Kinloch, I., Maynard, A., Wallace, W.A., Seaton, A., Stone, V., Brown, S.,  
642 Macnee, W., Donaldson, K., 2008. Carbon nanotubes introduced into the abdominal cavity of mice  
643 show asbestos-like pathogenicity in a pilot study. Nature nanotechnology 3, 423-428.

644 Praetorius, A., Scheringer, M., Hungerbühler, K., 2012. Development of environmental fate models  
645 for engineered nanoparticles - a case study of TiO<sub>2</sub> nanoparticles in the Rhine River. Environmental  
646 Science & Technology 46, 6705-6713.

647 Quik, J.T.K., de Klein, J.J.M., Koelmans, A.A., 2015. Spatially explicit fate modelling of nanomaterials in  
648 natural waters. Water research 80, 200-208.

649 Quik, J.T.K., Stuart, M.C., Wouterse, M., Peijnenburg, W., Hendriks, A.J., van de Meent, D., 2012.  
650 Natural colloids are the dominant factor in the sedimentation of nanoparticles. Environmental  
651 Toxicology and Chemistry 31, 1019-1022.

652 Quik, J.T.K., Velzeboer, I., Wouterse, M., Koelmans, A.A., Meent, D.v.d., 2014. Heteroaggregation and  
653 sedimentation rates for nanomaterials in natural waters. Water research 48, 169-179.

654 Raies, A.B., Bajic, V.B., 2016. In silico toxicology: computational methods for the prediction of  
655 chemical toxicity. Wiley Interdiscip Rev Comput Mol Sci 6, 147-172.

656 Reisfeld, B., Mayeno, A.N., 2012. What is Computational Toxicology? in: Reisfeld, B., Mayeno, A.N.  
657 (Eds.). Computational Toxicology: Volume I. Humana Press, Totowa, NJ, pp. 3-7.

658 Renard, K., Foster, G., Weesies, G., McCool, D., Yoder, D., coordinators, 1997. Predicting Soil Erosion  
659 by Water: A Guide to Conservation Planning with the Revised Universal Soil Loss Equation (RUSLE).  
660 in: Agriculture, U.S.D.o. (Ed.), p. 404.

661 RIVM. 2016. "ConsExpo nano version 2.0." from <http://www.consexponano.nl/>

662 Russell, W.M.S., Burch, R.L., 1959. The Principles of Humane Experimental Technique. Methuen,  
663 London.

664 Sabljčić, A., Güsten, H., Verhaar, H., Hermens, J., 1995. QSAR modelling of soil sorption. Improvements  
665 and systematics of log KOC vs. log KOW correlations. Chemosphere 31, 4489-4514.

666 Sanchez Jimenez, A., Varet, J., Poland, C., Fern, G.J., Hankin, S.M., van Tongeren, M., 2016. A  
667 comparison of control banding tools for nanomaterials. J Occup Environ Hyg 13, 936-949.

668 Sayes, C., Ivanov, I., 2010. Comparative study of predictive computational models for nanoparticle-  
669 induced cytotoxicity. Risk analysis : an official publication of the Society for Risk Analysis 30, 1723-  
670 1734.

671 Schneider, T., Brouwer, D.H., Koponen, I.K., Jensen, K.A., Fransman, W., Van Duuren-Stuurman, B.,  
672 Van Tongeren, M., Tielemans, E., 2011. Conceptual model for assessment of inhalation exposure to  
673 manufactured nanoparticles. J Expo Sci Environ Epidemiol 21, 450-463.

674 Schneider, T., Jensen, K.A., 2009. Relevance of aerosol dynamics and dustiness for personal exposure  
675 to manufactured nanoparticles. *Journal of Nanoparticle Research* 11, 1637-1650.

676 Schwabe, F., Schulin, R., Rupper, P., Rotzetter, A., Stark, W., Nowack, B., 2014. Dissolution and  
677 transformation of cerium oxide nanoparticles in plant growth media. *Journal of Nanoparticle*  
678 *Research* 16.

679 Shityakov, S., Roewer, N., Broscheit, J.-A., Förster, C., 2017. In silico models for nanotoxicity  
680 evaluation and prediction at the blood-brain barrier level: A mini-review. *Computational Toxicology*  
681 *2*, 20-27.

682 Singh, K.P., Gupta, S., 2014. Nano-QSAR modeling for predicting biological activity of diverse  
683 nanomaterials. *RSC Advances* 4, 13215.

684 Sizochenko, N., Leszczynski, J., 2017. Review of Current and Emerging Approaches for Quantitative  
685 Nanostructure-Activity Relationship Modeling. 1704-1721.

686 Sizochenko, N., Rasulev, B., Gajewicz, A., Kuz'min, V., Puzyn, T., Leszczynski, J., 2014. From basic  
687 physics to mechanisms of toxicity: the "liquid drop" approach applied to develop predictive  
688 classification models for toxicity of metal oxide nanoparticles. *Nanoscale* 6, 13986-13993.

689 Tamm, K., Sikk, L., Burk, J., Rallo, R., Pokhrel, S., Madler, L., Scott-Fordsmand, J.J., Burk, P., Tamm, T.,  
690 2016. Parametrization of nanoparticles: development of full-particle nanodescriptors. *Nanoscale* 8,  
691 16243-16250.

692 Tantra, R., Oksel, C., Puzyn, T., Wang, J., Robinson, K.N., Wang, X.Z., Ma, C.Y., Wilkins, T., 2014.  
693 Nano(Q)SAR: Challenges, pitfalls and perspectives. *Nanotoxicology*, 1-7.

694 Therezien, M., Thill, A., Wiesner, M.R., 2014. Importance of heterogeneous aggregation for NP fate in  
695 natural and engineered systems. *Science of The Total Environment* 485-486, 309-318.

696 Thomas, D.G., Gaheen, S., Harper, S.L., Fritts, M., Klaessig, F., Hahn-Dantona, E., Paik, D., Pan, S.,  
697 Stafford, G.A., Freund, E.T., Klemm, J.D., Baker, N.A., 2013. ISA-TAB-Nano: a specification for sharing  
698 nanomaterial research data in spreadsheet-based format. *BMC biotechnology* 13, 2.

699 Toropov, A.A., Leszczynska, D., Leszczynski, J., 2007. Predicting water solubility and octanol water  
700 partition coefficient for carbon nanotubes based on the chiral vector. *Computational biology and*  
701 *chemistry* 31, 127-128.

702 Toropova, A.P., Toropov, A.A., 2016. Assessment of nano-QSPR models of organic contaminant  
703 absorption by carbon nanotubes for ecological impact studies. *Materials Discovery* 4, 22-28.

704 Toropova, A.P., Toropov, A.A., Manganelli, S., Leone, C., Baderna, D., Benfenati, E., Fanelli, R., 2016.  
705 Quasi-SMILES as a tool to utilize eclectic data for predicting the behavior of nanomaterials.  
706 *NanoImpact* 1, 60-64.

707 Urbaszek, P., Gajewicz, A., Sikorska, C., Haranczyk, M., Puzyn, T., 2017. Modeling adsorption of  
708 brominated, chlorinated and mixed bromo/chloro-dibenzo-p-dioxins on C60 fullerene using Nano-  
709 QSPR. *Beilstein journal of nanotechnology* 8, 752-761.

710 van Kesteren, P.C., Cubadda, F., Bouwmeester, H., van Eijkeren, J.C., Dekkers, S., de Jong, W.H.,  
711 Oomen, A.G., 2015. Novel insights into the risk assessment of the nanomaterial synthetic amorphous  
712 silica, additive E551, in food. *Nanotoxicology* 9, 442-452.

713 Waalewijn-Kool, P.L., Diez Ortiz, M., van Straalen, N.M., van Gestel, C.A., 2013. Sorption, dissolution  
714 and pH determine the long-term equilibration and toxicity of coated and uncoated ZnO nanoparticles  
715 in soil. *Environmental pollution* 178, 59-64.

716 Wang, P., Keller, A.A., 2009. Natural and Engineered Nano and Colloidal Transport: Role of Zeta  
717 Potential in Prediction of Particle Deposition. *Langmuir : the ACS journal of surfaces and colloids* 25,  
718 6856-6862.

719 Wang, X., Zhang, L., Moran, M.D., 2010. Uncertainty assessment of current size-resolved  
720 parameterizations for below-cloud particle scavenging by rain. *Atmospheric Chemistry and Physics*  
721 *Discussions* 10, 2503-2548.

722 Westerhoff, P., Nowack, B., 2013. Searching for Global Descriptors of Engineered Nanomaterial Fate  
723 and Transport in the Environment. *Accounts of chemical research* 46, 844-853.

724 Winkler, D.A., 2016. Recent advances, and unresolved issues, in the application of computational  
725 modelling to the prediction of the biological effects of nanomaterials. *Toxicology and applied*  
726 *pharmacology* 299, 96-100.

727 Wyrzykowska, E., Mikołajczyk, A., Sikorska, C., Puzyn, T., 2016. Development of a novel in silico model  
728 of zeta potential for metal oxide nanoparticles: a nano-QSPR approach. *Nanotechnology* 27, 445702.

729 Zalk, D.M., Paik, S.Y., Swuste, P., 2009. Evaluating the Control Banding Nanotool: a qualitative risk  
730 assessment method for controlling nanoparticle exposures. *Journal of Nanoparticle Research* 11,  
731 1685-1704.

732

733